

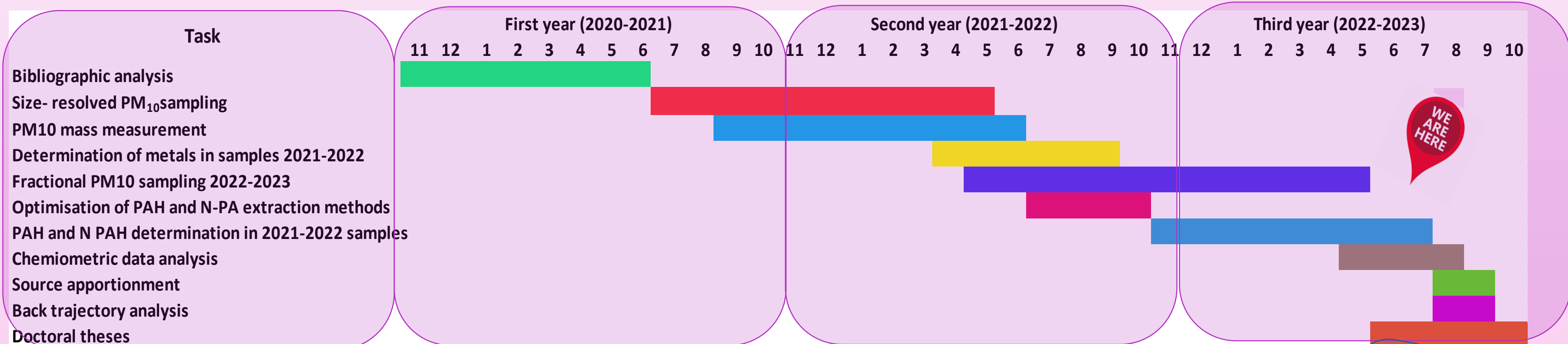


# “ORGANIC AND INORGANIC CONTAMINANTS OF ATMOSPHERIC PARTICULATE MATTER”

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## INTRODUCTION

The main objective of this PhD project is to sample, chemically characterize and study the spatiotemporal evolution of organic and inorganic pollutants in atmospheric particulate matter in the city of Ancona. In particular, the second year of the PhD project focused on: (1) the inorganic chemical characterization of the aerosol samples collected; (2) a second sampling campaign; (3) the assessment of the seasonal evolution of the aerosol's chemical composition; (4) the evaluation of the main sources of atmospheric pollutants.



## EXPERIMENTAL SECTION

### First sampling campaign (July 2021 - March 2022)

#### Sample treatment

Airborne samples of the first sampling campaign were subjected to microwave – assisted digestion (MARS 6) in closed PTFE vessel by adding 5 ml of HNO<sub>3</sub> + 1 ml of HF + 1 ml of H<sub>2</sub>O<sub>2</sub>



#### Quantification of metals

Metal concentrations (Al, As, Cd, Cr, Cu, Fe, Mn, Ni, Pb, V) were determined by Atomic Absorption with graphite furnace (GF-AAS), while particulate Hg was measured by the Direct Mercury Analyzer (DMA)



#### Quality control

Field blanks were collected at the beginning and at the end of the field campaign.

Accuracy was assessed by analysing the certified reference material NIST 1648a. Results are in agreement with the certified reference values.

### Università Politecnica of Marche Campus



Flux : 1.13 m<sup>3</sup> min<sup>-1</sup>  
 Sampling strategy  
 Summer: 3 days twice a month  
 Rest of the year: 3 days once a month

### Second sampling campaign (June 2022 - September 2022)

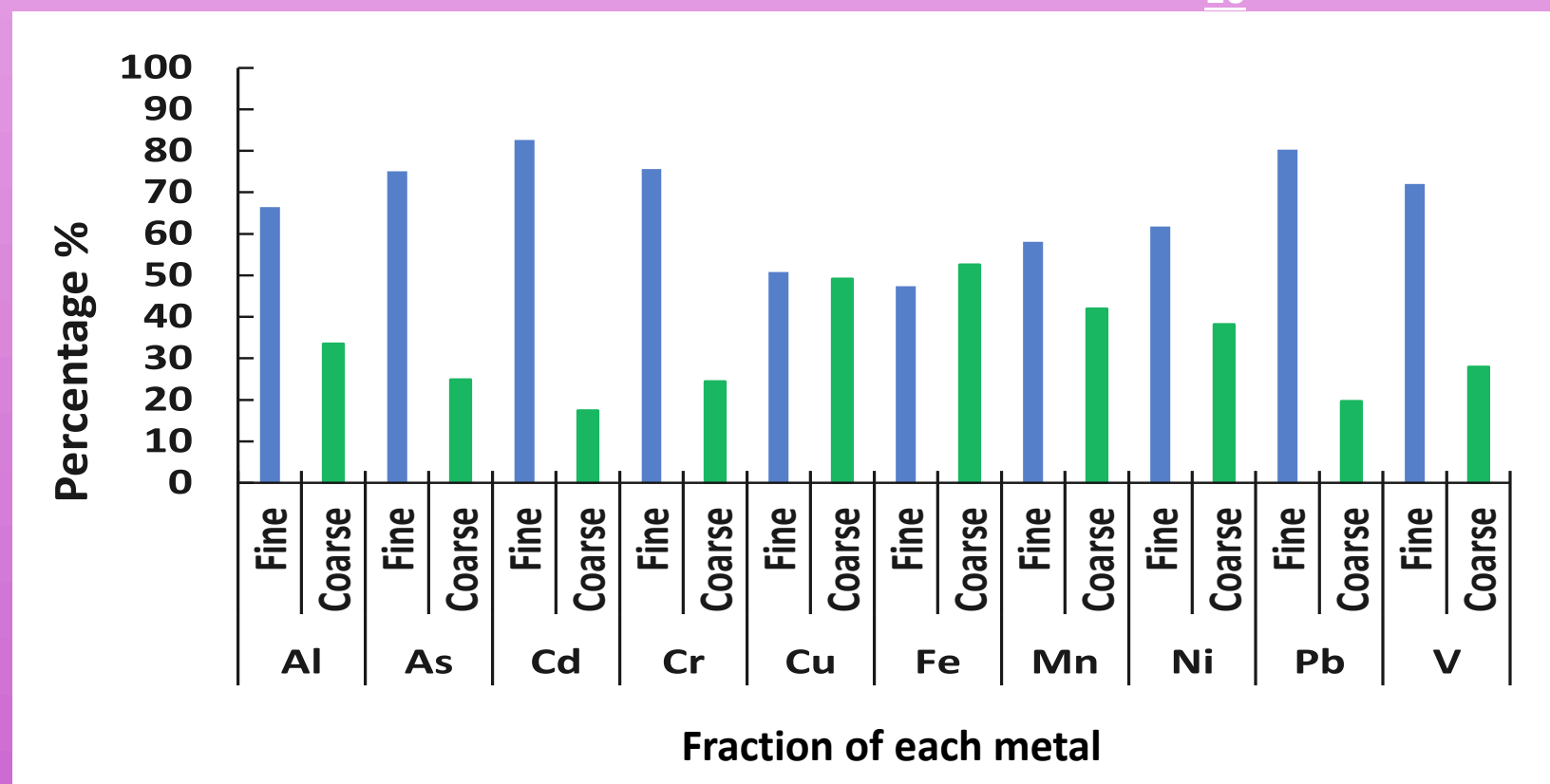
Same sampling strategy of the first campaign. Quartz fibre filters were used for the determination of Polycyclic Aromatic Hydrocarbons (PAHs) and nitro-derived PAH

## RESULTS AND DISCUSSION

### Mean concentration of metals in PM<sub>10</sub>

- Al = 491 ± 8 ng/m<sup>3</sup>
- Fe = 478 ± 9 ng/m<sup>3</sup>
- Mn = 5.6 ± 0.1 ng/m<sup>3</sup>
- As = 322 ± 6 pg/m<sup>3</sup>
- Cd = 56 ± 1 pg/m<sup>3</sup>
- Cr = 1.42 ± 0.03 ng/m<sup>3</sup>
- Cu = 4.36 ± 0.08 ng/m<sup>3</sup>
- Ni = 1.35 ± 0.02 ng/m<sup>3</sup>
- Pb = 1.84 ± 0.03 ng/m<sup>3</sup>
- V = 1.70 ± 0.04 ng/m<sup>3</sup>

### Size distribution of elements in PM<sub>10</sub>



Fine fraction: D<sub>p</sub> < 2.5 μm  
 Coarse fraction: 2.5 μm < D<sub>p</sub> < 10 μm

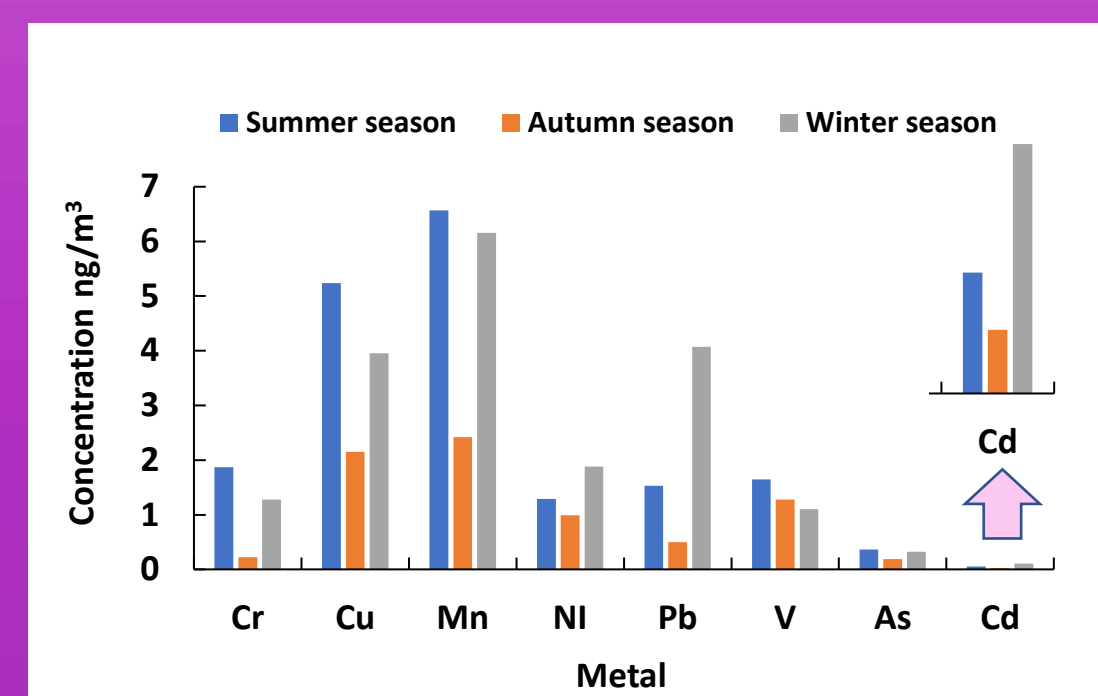
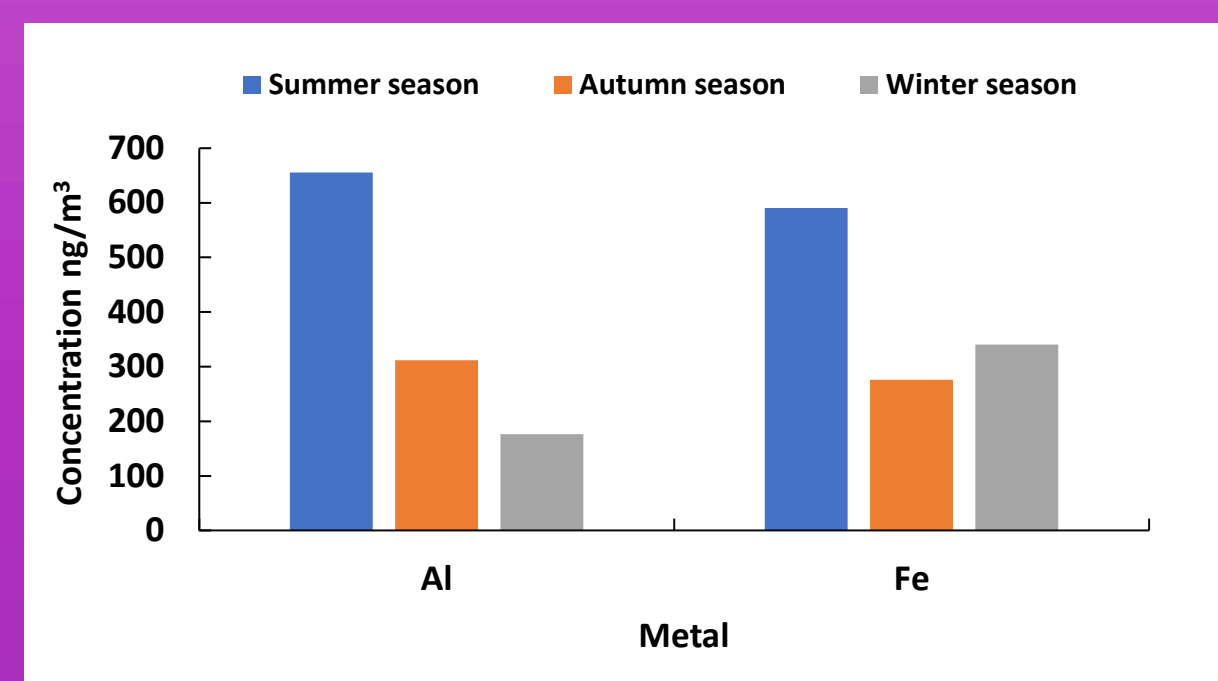
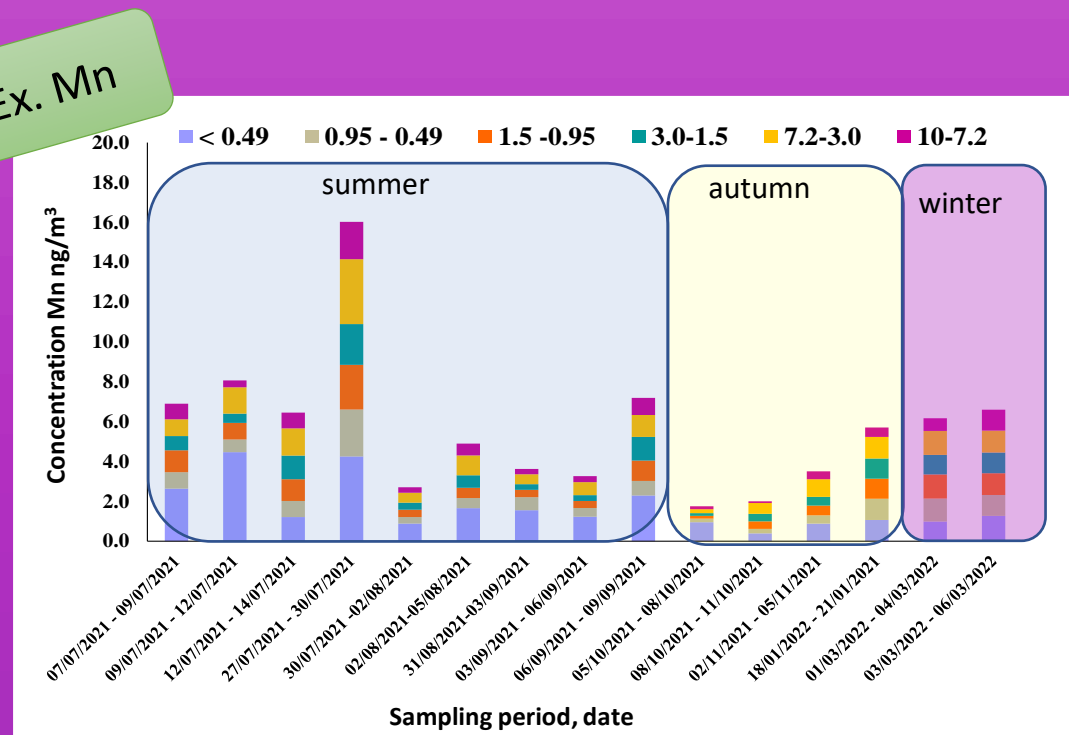
Al, As, Cd, Cr, Mn, Ni, Pb, V → Mainly present in the fine fraction  
 Cu, Fe → Almost equally distributed between the fine and the coarse fractions

### Dlgs n. 155/2010

Metals	Annual average conc. (ng/m <sup>3</sup> )
As	6
Cd	5
Ni	20
Pb	500

As, Ni: 10 times below legal limit  
 Cd: 100 times below legal limit  
 Pb: 250 times below legal limit

### Seasonal evolution of the chemical composition of PM<sub>10</sub>



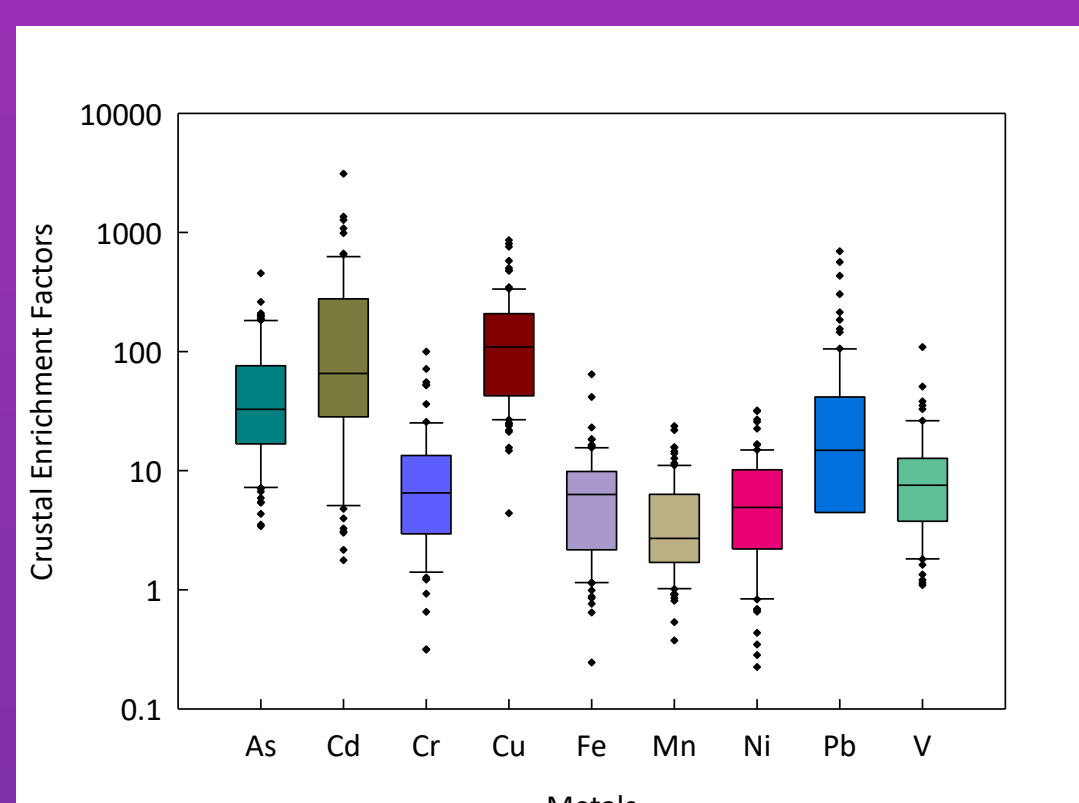
### Source apportionment

Enrichment factor (EF) calculation is a first step in the source apportionment evaluation to differentiate the possible sources of elements in the atmospheric particulate matter<sup>1</sup>.

$$EF_i = \frac{(C_i/C_{Al})_{PM}}{(C_i/C_{Al})_{UCC}}$$

(C<sub>i</sub>/C<sub>Al</sub>)<sub>PM</sub> = concentration ratio of the element *i* over the reference element Al in the atmospheric particulate matter

(C<sub>i</sub>/C<sub>Al</sub>)<sub>UCC</sub> = the abundance ratio of the element *i* over the reference element Al in the upper continental crust (UCC)<sup>2</sup>.



EFs ≤ 10: Fe, Mn, Cr, Ni, V → GEOGENIC ORIGIN  
 10 < EFs < 100: As, Pb → MODERATELY ENRICHED  
 EFs > 100: Cd, Cu → GREATLY ENRICHED

### FUTURE PERSPECTIVES

- Set up of the analytical extraction procedure for PAHs and nitro-PAHs in atmospheric aerosol
- Determination of PAHs and n-PAHs in samples collected during the second sampling campaign
- Chemometric treatment of data
- Air mass backtrajectories model to determine the sources of elements
- Source apportionment

### REFERENCES

1. Zoller, W.H., Gladney, E.S., Duce, R.A. Atmospheric concentrations and sources of trace metals at the South Pole. Science 1974, 183, 198–201  
 2. Wedepohl, K. Geochim. Cosmochim. Acta 1995, 59, 1217–1232.